

Bohmian Mechanics:

A Trajectory Picture of Quantum Mechanics

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Why trajectory pictures of quantum mechanics?

Fundamental pictures of quantum mechanics:

- Heisenberg (1925) \Rightarrow Operators ("black box")
- Schödinger (1926) \Rightarrow Deterministic wave fields
- Feynman (1948) \Rightarrow Classical-like paths and waves

Quantum system = wave





Demonstration of single-electron buildup of an interference pattern

Tonomura, Endo, Matsuda, Kawasaki and Ezawa, *Am. J. Phys.* **57**, 117 (1989).







Particle distributions behave as waves ...

... but individual particles behave as individual point-like particles!

Explaining both behaviors within the same theoretical framework is precisely the reason why trajectory pictures of quantum mechanics are needed or desirable



- 1) The evolution of the wave function, $\Psi(\vec{r},t)$, is given by the timedependent Schrödinger equation.
- 2) The particle momentum is determined by Jacobi's law, $\vec{p} = \nabla S$, obtaining the Bohmian trajectories by integrating this law of motion.
- 3) There is no any prediction or control a priori over the particle initial position, but rather some statistical information about it given by the probability density, $\rho(\vec{r}) = |\Psi(\vec{r})|^2$.



$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \Psi + V \Psi$$

$$\Psi = R e^{iS\hbar}$$
$$\left\{ \begin{array}{c} \frac{\partial S}{\partial t} + \frac{(\nabla S)^2}{2m} + V - \frac{\hbar^2}{2m} \frac{\nabla^2 R}{R} = 0\\ \frac{\partial R^2}{\partial t} + \nabla \cdot \left(R^2 \frac{\nabla S}{m}\right) = 0 \end{array} \right\}$$

$$\begin{cases} i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \Psi + V \Psi \\ \vec{p} = \nabla S \end{cases}$$







> Mathematical foundations and implied philosophy

- P.R. Holland The Quantum Theory of Motion (1993)
- D. Dürr Bohmsche Mechanik (2001)

> Development of trajectory-based algorithms

R.E. Wyatt – Quantum Dynamics with Trajectories (2005)

Reinterpretation of quantum phenomena

A.S. Sanz, S. Miret-Artés – Trajectory Pictures of Quantum Mechanics (2009?)



- Diffraction by slit gratings
- Atom-surface scattering
- Fractal Bohmian mechanics
- Beam interference and interferometry



1.- Dynamical characterization of optical/quantum regions





2.- Reproduction of phenomena and/or effects as in the experiment





3.- Contextuality: Particles only cross one slit; the wave, both





Applications: Multi-slit diffraction









4.- Initial and final regions of configuration (position) space can be unambiguously related





5.- Quantum particles are affected by a sort of quantum pressure



Soft (realistic) two-slit potential





Applications: Two-slit diffraction





Applications: He-CO/Cu elastic scattering



6.- Detection of quantum vortices







7.- Causal explanation and characterization of the *Talbot effect*

Periodicity in x:

d

Periodicity in z:









quantum carpet



Applications: The Talbot effect



Talbot structure



Applications: The Talbot effect









Applications: Fractal Bohmian mechanics





9.- Generalization of Bohmian mechanics to deal with *fractal* quantum states

Fractal quantum dynamics:

1





Figure 2. (a) QF-trajectories associated with a highly delocalized particle in a box. (b) Measure of the fractal dimension of a sample of QF-trajectories with initial positions: $x_0 = 0.01$ (**I**), $x_0 = 0.1$ (**I**), $x_0 = 0.49$ (**I**), $x_0 = 0.499$ (**I**), and $x_0 = 0.5$ (Δ).



Applications: Interference and interferometry







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Applications: Interference and interferometry







Bohmian mechanics provides a robust and consistent framework to analyze and understand the dynamical behavior of quantum systems, which allows to treat particles as in classical mechanics (i.e., as individual entities) and, at the same time, to observe the well-known wave-like behaviors characteristic of the standard version of quantum mechanics.

In other words, Bohmian mechanics can be an important tool to create the quantum intuition necessary to think the quantum world.



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