Light induced atomic desorption [LIAD] and related phenomena





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In the past atom-surface interaction [no light!] was a crucial problem

In optical pumping atom-wall collisions destroy spin orientation

In light induced drift atoms adsorbed at the surface make drift velocity very slow surface density in fact grows up with the adsorption energy Solution --> walls coating with organic film

New momentum grew up because of new experiments made with ultra-thin cells and nano structured materials, because of miniaturization of many apparatus (traps, atom chips etc.) and the discovery of new phenomena.

+ light !

It is well known that Light hitting a surface triggers many different processes depending on its intensity, frequency, pulsed or c.w. regime. High intensity produces ionization, plasma formation, ablation etc.

+Low intensity

Weak light triggers other interesting effects with alkali atoms in coated cell and nanoporous silica.



Light Induced Atomic Desorption





Few isolated atoms at the surface



LIAD Effect





Fig. 18. – Transmission of a coated cell at room temperature. The laser frequency is kept on resonance with the Rb atoms. 1 corresponds to the cell transmission in the dark; 2: transmission when the ceiling light is on; 3: flash-lamp + ceiling light on.



S.N.Atutov et al. PRA 60 (1999) 4693 C. Marinelli et al. Eur. Phys. J. D13 (2001) 231 C. Marinelli et al. Eur/ Phys. J. D 37 (2006) 319

Similar one dimension model has been proposed by Budker et al. for paraffin. PRA 66 (2002) 042903 with essentially the same results.



Surface + bulk effect Light modifies the atom diffusion inside the coating. The coating becomes an atomic reservoir.





Budker et al. PRA 66 (2002) 042903

" there are several proposals to explain the dependence of LIAD on the frequency of the desorbing light. Bonch-Bruevich et al. suggest that the frequency dependence is related to the absorption spectrum of adsorbed atoms. Another idea is that the alkali-metal atoms form quasimolecular bonds with the coating and the photon energy must exceed some threshold. It is still difficult to distinguish between these various possibilities with our present data".

pulsed excitation --> high T -

"cw" weak illumination --> diffusion - no dependence on atom

Different regimes different mechanisms -

PDMS film quite complex system and diffusion goes on in the dark!



Desorption dynamics of Rb and Cs in the same cell





C. Marinelli et al., Eur.Phys.J. D 37 (2006)

Image: Second polymers



500

××

2.4

×

2.8

*

C. Marinelli et al., Eur.Phys.J. D 37 (2006)



LAD from An explanation of polymers LIAD from PDMS Evidences

low desorption energy

increase of desorption efficiency with photon energy

no resonant behavior or threshold with desorbing photon energy desorption characteristics independent from the alkali species

Non resonant inelastic light scattering

Desorbing light interacts with the polymer Excitation of polymer phonons via Raman and Brillouin scattering

Increase of the atomic diffusion and desorption probability



Light off

Light on





Rb source @T<0°C Rb source at room temperature





We use as host matrix for atoms and nanoparticles porous silica with a mean pore diameter of 17 nm and a free volume of about 50% of the whole silica mass.

Pore volume: 500mm³/g Pore surface: 100m²/g Mean pore diameter: 17nm SiO₂>96%



~17nm





The porous glass sample is a rectangular plate $30 \times 15 \times 1 \text{ mm}^3$ in size. It is placed inside a Pyrex resonance cell, kept at room temperature, and filled with rubidium or cesium.





Optics Express 16, 1377 (2008) Phys. Rev.Lett. 97 157404 (2006)



The desorbing rate R and the maximum increase of the vapour density δ_{max} are measured as function of the desorbing light intensity and frequency.



Photodesorption dependence on light intensity

Photo-induced processes in 17nm porous glass

× Rb • Cs

×

1.6

×

Desorbing photon energy (eV)

2.0

4

2

1

 δ^{max} 3



* [×]

2.8

××

2.4

The decay time depends on the desorbing light intensity, wavelength and on the illumination time



A. Burchianti et al. Physical Review Letters 97 (2006)



NIR-BLUE-NIR sequence of colours







Six shots

By alternately illuminating the sample with blue-green and NIR light we find that the PG sample remembers the illumination sequence. Sequence light pulses: red+green+red+2green+red



260mW@810nm/16mm²

Storing and erasing images in Rb loaded PG

Optics Express 16, 1377 (2008)



20mW/cm² at 2.3eV; 2min



2.5W/cm² at 1.5eV; 30s

In the dark the system relaxes to the equilibrium condition





Cluster growth induced by UV-visible light is correlated to LIAD

NIR light induce both cluster evaporation via SPID and cluster growth via LIAD





Cs adsorbed on porous glass

Photo-induced

processes in



Change of PG Absorbance Cluster Formation

The photon energy dependence of **Surface Plasmon Induced Desorption** is dominated by dipolar surface plasmon frequency. For a spherical metal particle we get





we apply Gans theory that describes the optical properties of randomly oriented spheroids with size R << λ

$$\sigma_{ext}(\omega) = \frac{\omega}{3c} \cdot V \cdot \varepsilon_m^{3/2} \cdot \sum_i \frac{\varepsilon_2(\omega)(1/P_i^2)}{\varepsilon_2(\omega)^2 + \left\{\varepsilon_1(\omega) + \varepsilon_m \frac{1-P_i}{P_i}\right\}^2}$$

$$P_c = \frac{1+e^2}{e^3} \cdot \left(e - \tan e^{-1}\right)$$

$$P_a = P_b = \frac{1}{2} \cdot (1-P_c) \qquad \text{for oblate} \\ \text{particles (c

$$e = \sqrt{\frac{a^2}{c^2} - 1}$$$$









R=3nm; c/a=0.8





Light controlled atomic dispensers - fast and clean

Atom delivery in nanostructures control of optical thicknes with light

Control of cluster formation in nanostructured materials

High vapor densities at room or lower temperatures

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